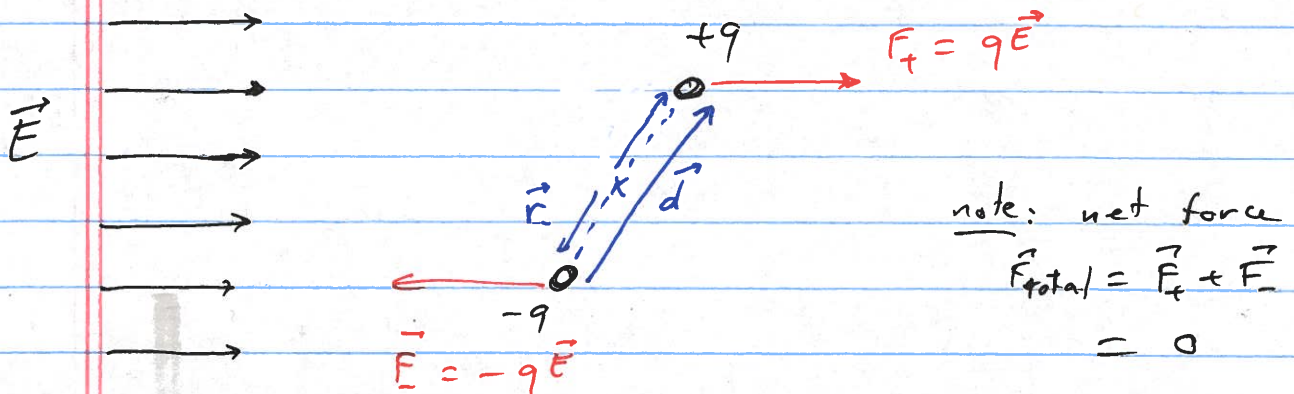


Wednesday, March 29, 2023

## Dipole Forces [chpt 4.1.3]

A) Torque on a dipole in a uniform E-field



$$\text{Torque: } \vec{\tau} = \vec{r}_+ \times \vec{F}_+ + \vec{r}_- \times \vec{F}_-$$

$$= q\vec{d} \times \vec{E}$$

↓ some algebra (short)

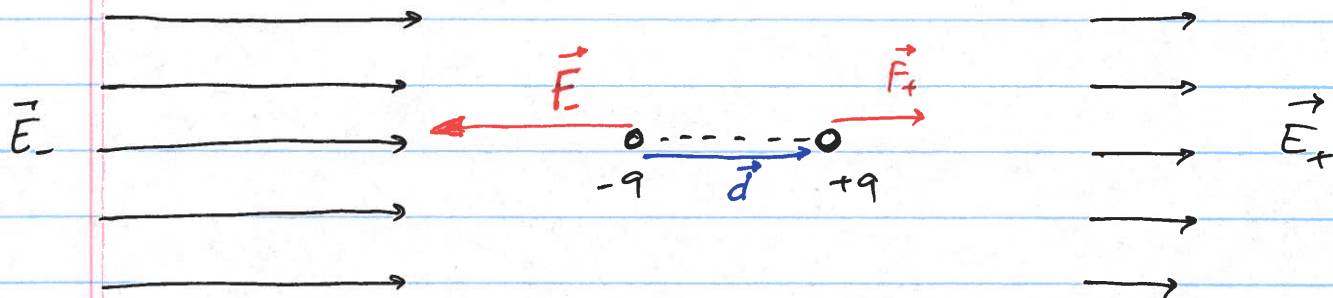
$$\Rightarrow \boxed{\vec{\tau} = \vec{p} \times \vec{E}} = \text{torque on a dipole}$$

note: torque is zero when  $\vec{p} \parallel \vec{E}$  (parallel or anti-parallel)  
stable unstable

↳ a molecule with a permanent dipole moment will align with the electric field.

↓  
 i.e. a polar molecule  
 ↳ e.g. water

B) Force on a dipole in a non-uniform E-field



$$\vec{F}_{total} = \vec{F}_+ + \vec{F}_- = q(\vec{E}_+ - \vec{E}_-)$$

note:  $(\vec{d} \cdot \vec{\nabla}) \vec{E} = (d_x \frac{\partial}{\partial x} + d_y \frac{\partial}{\partial y} + d_z \frac{\partial}{\partial z}) \vec{E}$

$\Delta \vec{E} = \dots = (\vec{d} \cdot \vec{\nabla}) \vec{E}$  some algebra

$$= (\vec{p} \cdot \vec{\nabla}) \vec{E}$$

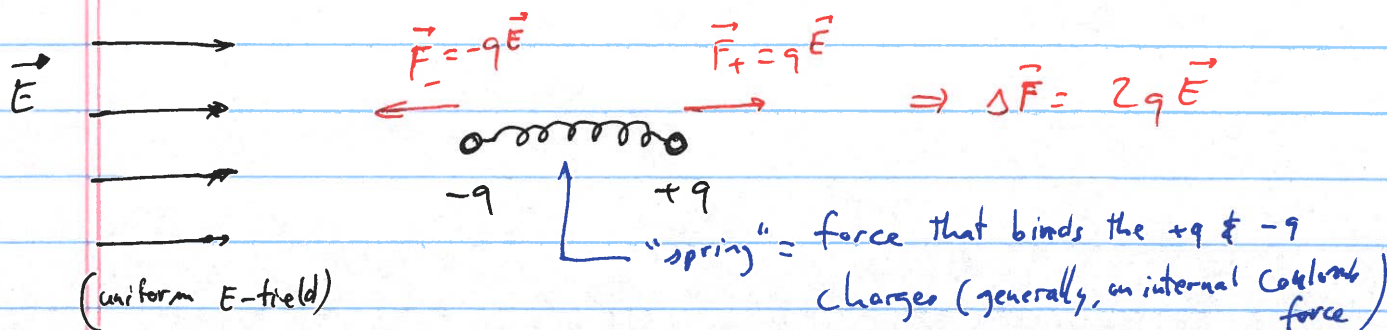
$\Rightarrow \boxed{\vec{F} = (\vec{p} \cdot \vec{\nabla}) \vec{E}}$  = force on a dipole in a non-uniform E-field.

note: if  $\vec{p} = \text{cst}$  (i.e. does not change orientation or magnitude) then  $\vec{F} = \vec{\nabla}(\vec{p} \cdot \vec{E})$

↳ The potential energy  $U$  of the dipole is then obtained from

$$\vec{F}_{dipole} = -\vec{\nabla} U_{dipole} \Rightarrow \boxed{U_{dipole} = -\vec{p} \cdot \vec{E}}$$

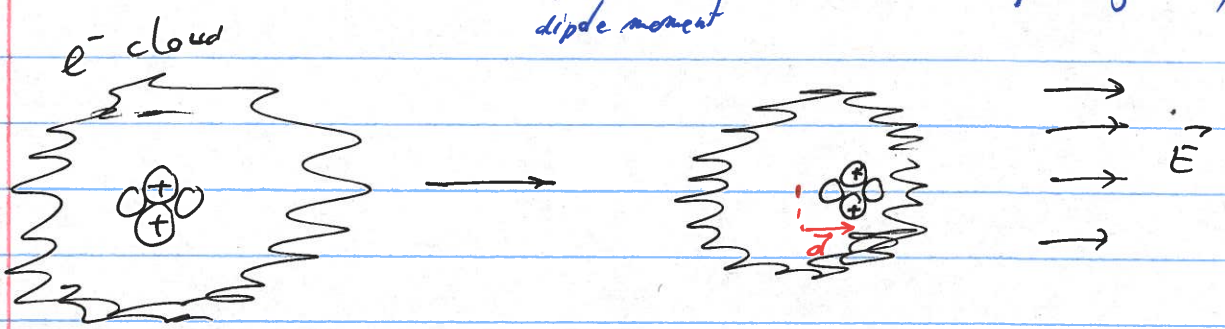
C) Internal force on a dipole, atom, molecule: Induced dipole moment  
[chpt. 4.1.2]



Internal "spring" displacement:  $\vec{F} = -k \Delta \vec{x}$

" $\vec{d}$ "  
gives rise to a dipole moment  $q \Delta \vec{x} = q \vec{d}$

Typically, for atoms & molecules,  $\vec{p} = \alpha \vec{E}$   
induced dipole moment      applied E-field      atomic polarizability



$\alpha = \text{DC polarizability} > 0$  [see Stark effect in QM]

Back to dipole force:  $\vec{F} = (\vec{p} \cdot \nabla) \vec{E} = (\alpha \vec{E} \cdot \nabla) \vec{E} = \nabla \left( \frac{1}{2} \alpha \vec{E}^2 \right)$

$\Rightarrow$  potential energy:  $U_{\text{atom}} = -\frac{1}{2} \alpha \vec{E}^2$

Energy of atom becomes more negative as the E-field increases.

$\Rightarrow$  atoms are attracted to regions of high E-field (and molecules)

$\Rightarrow$  atoms & molecules are high-field seekers (E-field)

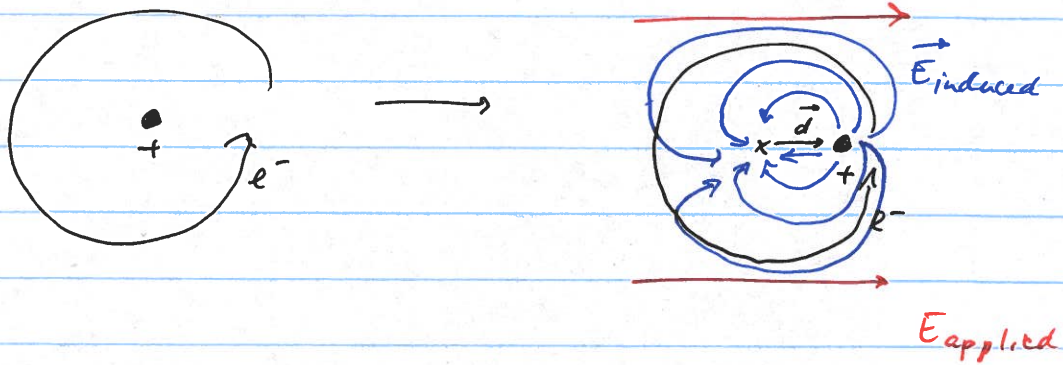
Example: For an IR laser, far from any atomic resonances, the laser is quasistatic to an atom ( $e^-$  dynamics/response is much faster than  $f_{\text{laser}} = \text{laser frequency}$ )

$\hookrightarrow$  atoms are attracted to the laser focus (where  $E_{\text{laser}}$  is highest)   
 high intensity  $\downarrow$



## Electrostatics in Matter

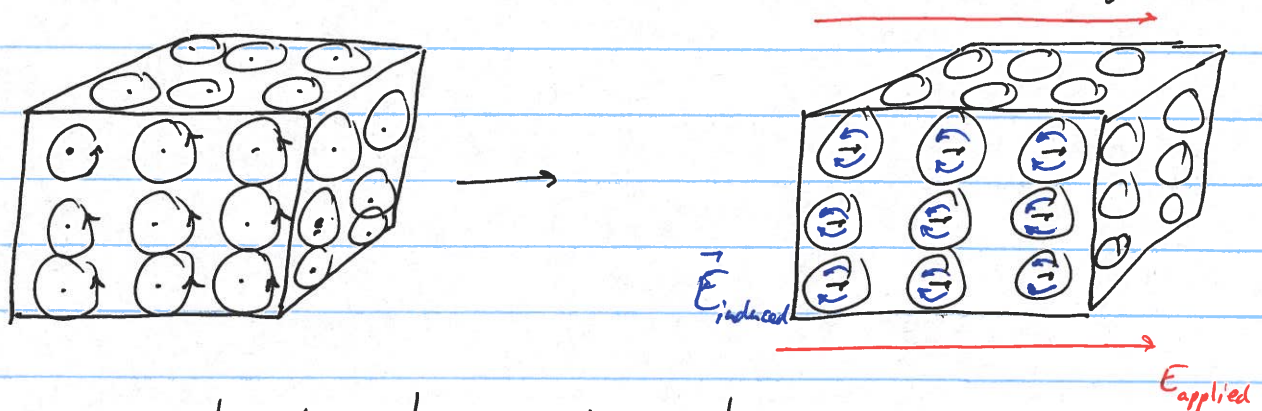
1) Consider a classical atom in an E-field:



$$\vec{P}_{\text{induced}} = \alpha \vec{E}$$

$\Rightarrow$  The induced dipole  $\vec{P}_{\text{induced}}$  produces its own E-field,  $\vec{E}_{\text{induced}}$ , which partially counteracts the applied E-field,  $\vec{E}_{\text{applied}}$ .

2) Consider a block of matter made up of similar-behaving atoms:



$\Rightarrow$  you expect to polarize the material and so there should be an associated  $\vec{E}_{\text{induced}}$

$\hookrightarrow$  dipole moment per unit volume:  $\vec{P} = \alpha \vec{E}$

we'll come back to this later

Thus  $\vec{E}_{\text{total}}$  is given by  $\vec{E}_{\text{total}} = \vec{E}_{\text{applied}} + \underbrace{\vec{E}_{\text{induced}}}_{\vec{E}_{\text{self}} \text{ or } \vec{E}_{\text{internal}}}$

→ you expect the charges in the material to not exactly cancel (i.e. you get local dipoles)

↳  $\left\{ \begin{array}{l} \text{Volume bound charge } \rho_b \\ \text{Surface bound charge } \sigma_b \end{array} \right.$

↳ total charge density:  $\rho_{\text{total}}(\vec{r}) = \rho_{\text{applied}}(\vec{r}) + \rho_b(\vec{r}) + \sigma_b(\vec{r})$

$\rho_{\text{free}} =$  free charge  
= charge put there by the experimentalist

We also require  $\int_{V \text{ of material}} \rho_b(\vec{r}) d^3r + \int_{S \text{ of material}} \sigma_b(\vec{r}) ds = 0 = \text{total bound charge}$

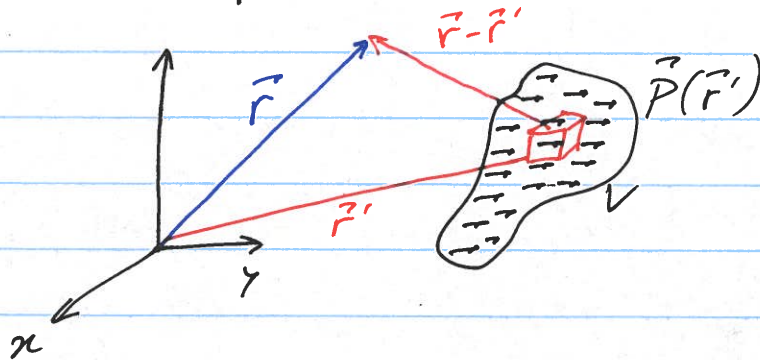
definition:  $\vec{P} \equiv$  dipole moment per unit volume

↳ we will model polarized matter as a collection of dipoles (no quadrupoles)

Q: What is the relationship between  $\vec{P}$ ,  $\rho_b$ , and  $\sigma_b$ ?

A: Let's derive it!

Consider the potential  $V(\vec{r})$  produced by a polarized material:



For a single dipole  $\vec{p}$ :  $V(\vec{r}) = \frac{1}{4\pi\epsilon_0} \frac{\vec{p} \cdot \hat{r}}{r^2}$

For a continuous distribution of  $\vec{p}$ 's:  $V(\vec{r}) = \frac{1}{4\pi\epsilon_0} \int_V \frac{\vec{P}(\vec{r}') \cdot (\vec{r} - \vec{r}')}{|\vec{r} - \vec{r}'|^2} d^3r'$

stopped here (i.e. our material)

However, we note that  $\vec{\nabla}_{r'} \left( \frac{1}{|\vec{r} - \vec{r}'|} \right) = \frac{(\vec{r} - \vec{r}')}{|\vec{r} - \vec{r}'|^2}$

a few lines of algebra

Thus  $V(\vec{r}) = \frac{1}{4\pi\epsilon_0} \int_V \vec{P}(\vec{r}') \cdot \vec{\nabla}_{r'} \left( \frac{1}{|\vec{r} - \vec{r}'|} \right) d^3r'$

$$= \frac{1}{4\pi\epsilon_0} \left[ \int_V \vec{\nabla} \cdot \left( \frac{\vec{P}(\vec{r}')}{|\vec{r} - \vec{r}'|} \right) d^3r' + \int_V \frac{(-\vec{\nabla} \cdot \vec{P})}{|\vec{r} - \vec{r}'|} d^3r' \right]$$

divergence theorem

$$\Rightarrow V(\vec{r}) = \frac{1}{4\pi\epsilon_0} \int_S \frac{\vec{P}(\vec{r}') \cdot \hat{n}'}{|\vec{r} - \vec{r}'|} ds' + \frac{1}{4\pi\epsilon_0} \int_V \frac{(-\vec{\nabla}_{r'} \cdot \vec{P}(\vec{r}'))}{|\vec{r} - \vec{r}'|} d^3r'$$

surface charge distribution

volume charge distribution



Thus we infer (or define):

volume bound charge:  $\rho_b(\vec{r}) \equiv -\vec{\nabla} \cdot \vec{P}(\vec{r})$

~~surface~~

surface bound charge:  $\sigma_b(\vec{r}) \equiv \vec{P}(\vec{r}) \cdot \hat{n}$

where  $\hat{n}$  = normal  
vector  
pointing out of  
surface