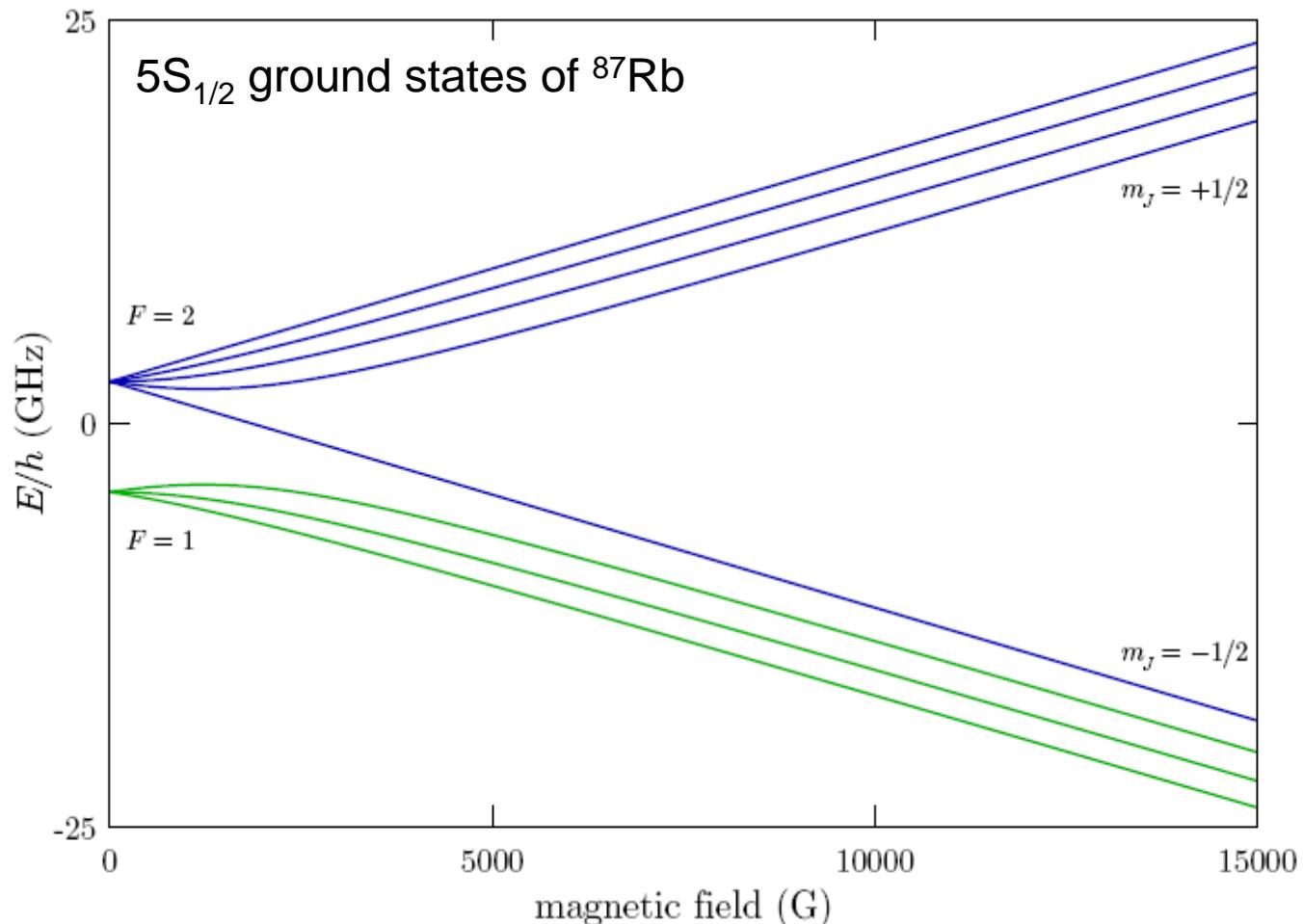
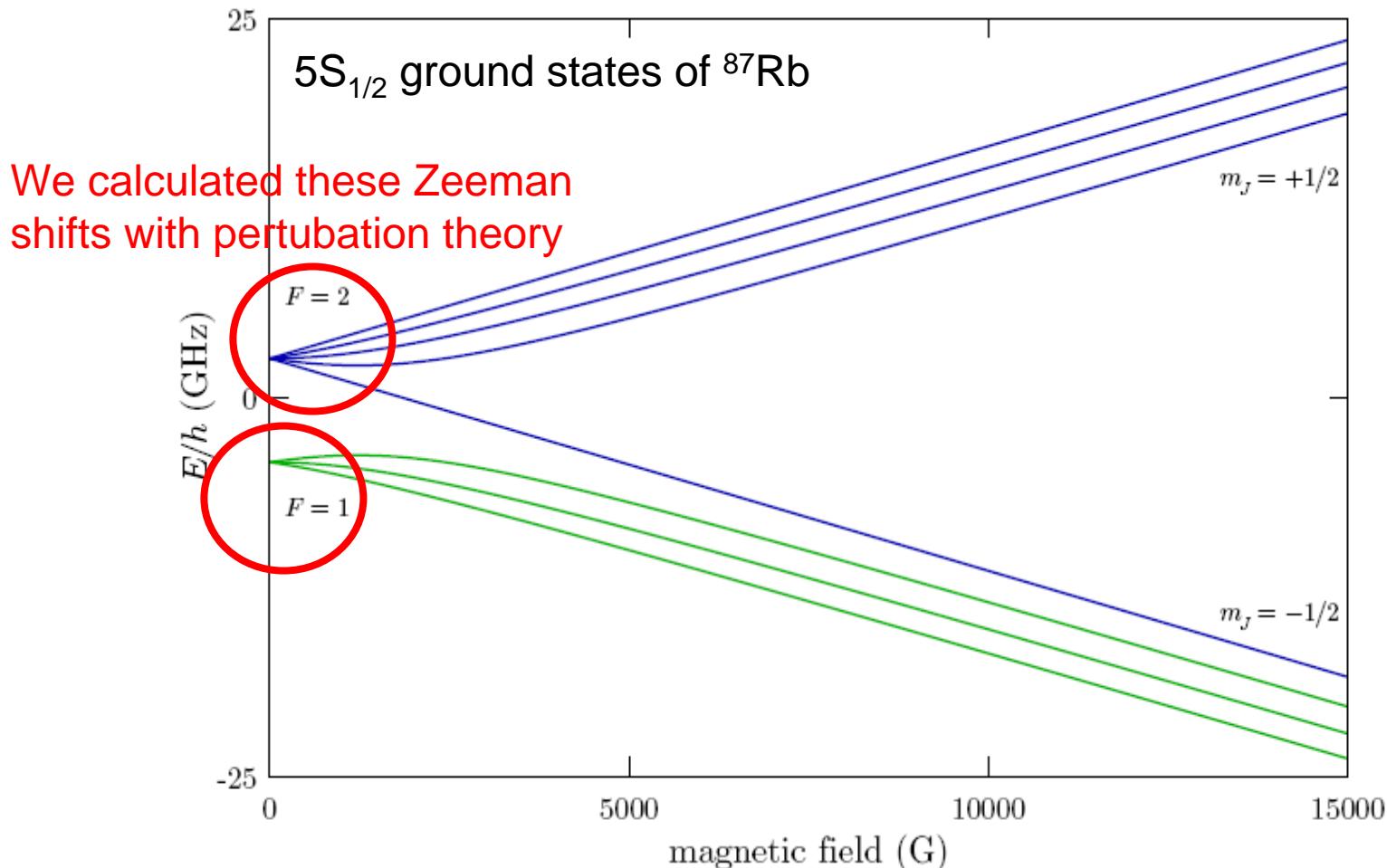


Zeeman Sub-Structure at High B-field



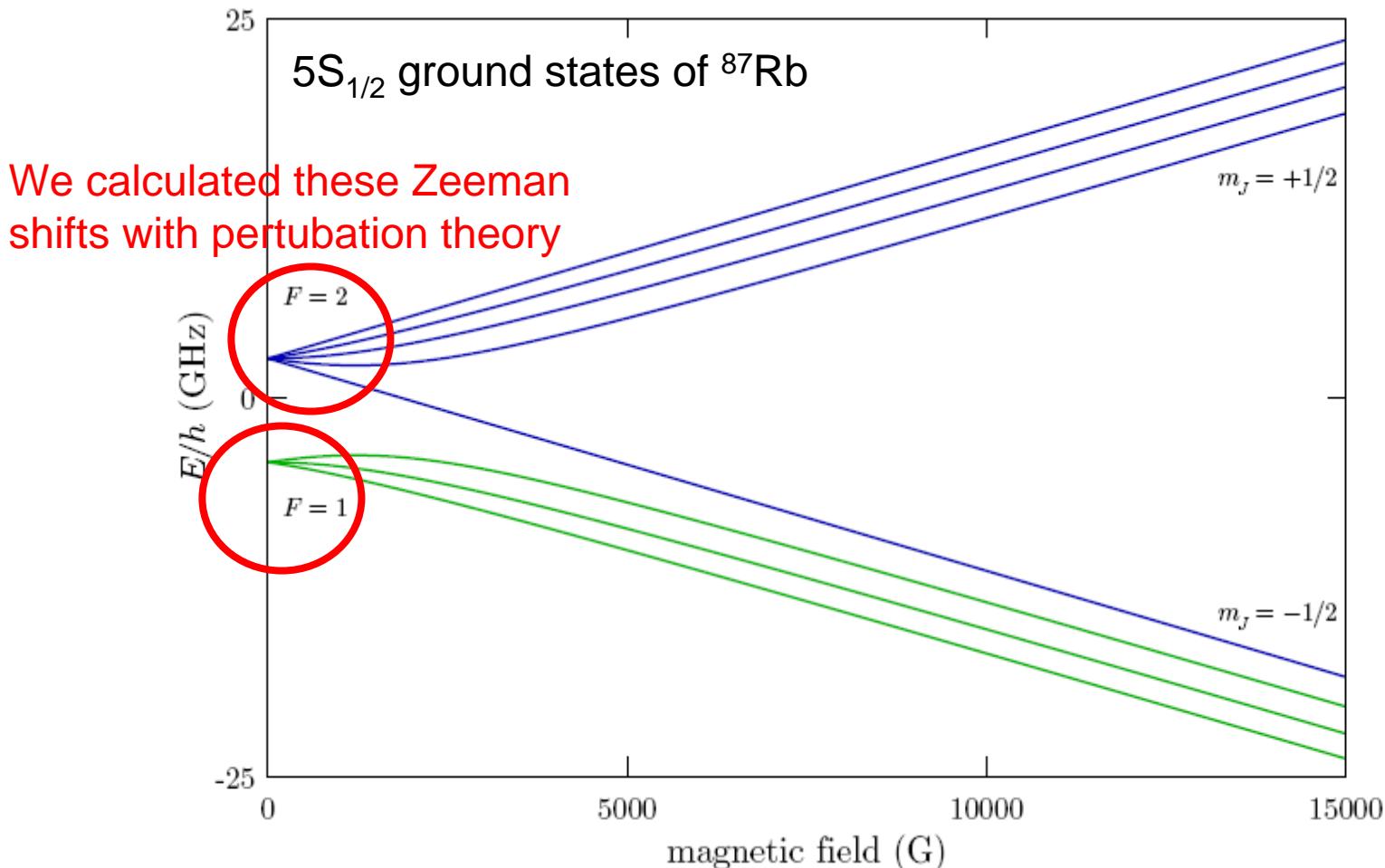
[Figure adapted from steck.us by Prof. Dan Steck, U. of Oregon (2010)]

Zeeman Sub-Structure at High B-field



Zeeman Sub-Structure at High B-field

How do you calculate this entire plot?



Zeeman shifts and the Hyperfine Hamiltonian

In the ground state (S state, so L=0):

$$H = H_0 + H_{\text{FineStructure}} + H_{\text{Hyperfine}} + H_{\text{Zeeman}}$$

Zeeman shifts and the Hyperfine Hamiltonian

In the ground state (S state, so L=0):

$$H = H_0 + H_{\text{FineStructure}} + H_{\text{Hyperfine}} + H_{\text{Zeeman}}$$

$$H = \frac{P^2}{2m} - \frac{e^2}{R} + \frac{e^2}{mc^2} \frac{1}{R^3} (\vec{L} \cdot \vec{S}) + hA(\vec{I} \cdot \vec{J}) + \underbrace{\frac{\mu_B}{\hbar} (g_s \vec{S} + g_I \vec{I})}_{\text{perturbation}}$$

Breit-Rabi Formula

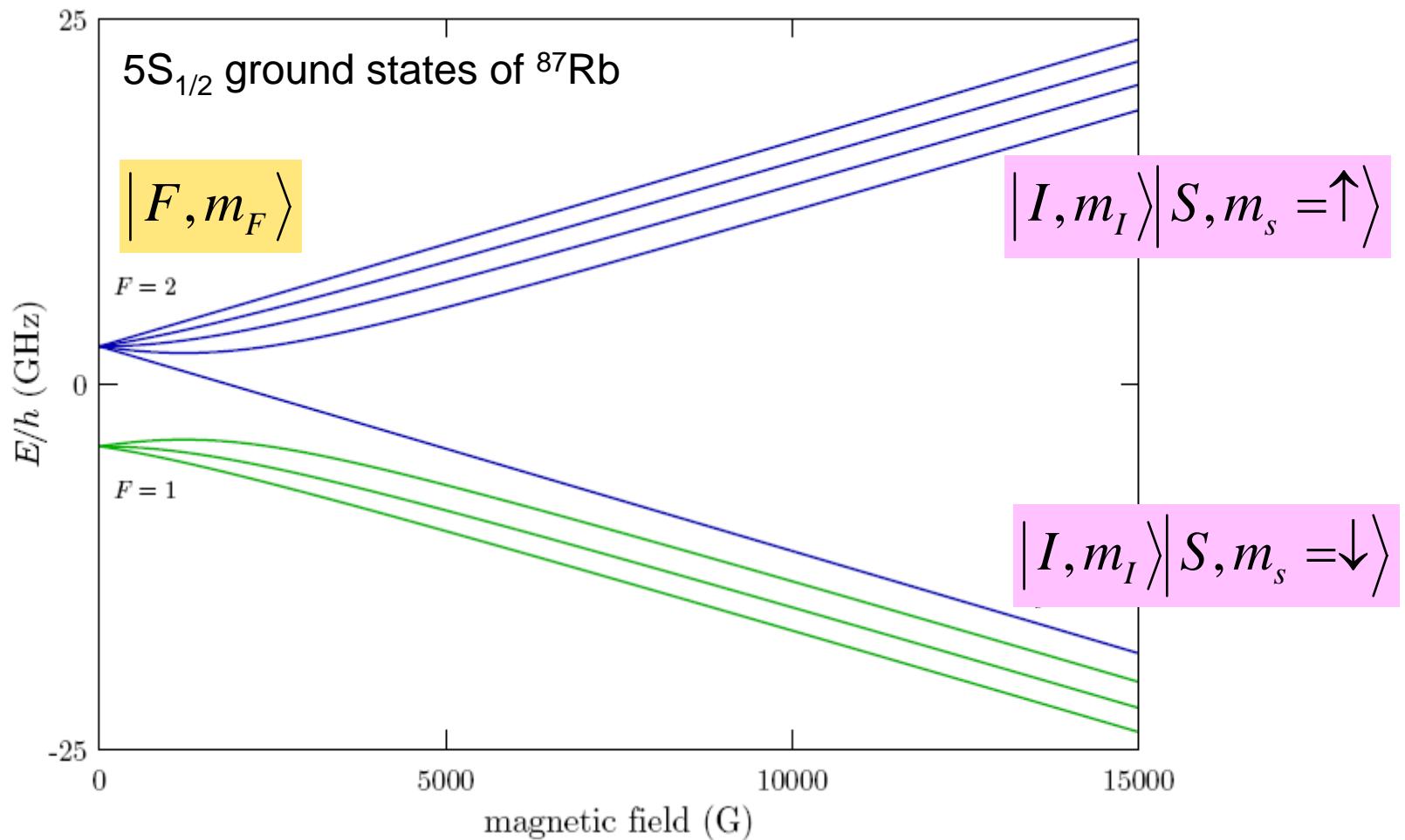
The Breit-Rabi formula for the Zeeman shift of atomic ground states is given by:

$$U(m_F, B) = g_I \mu_B m_F B + \frac{E_{hfs}}{2} \left(\pm \left(1 + \frac{4m_F x}{2I+1} + x^2 \right)^{1/2} - \frac{1}{2I+1} \right),$$

where the \pm is used for the $F = I \pm J$ state, respectively, and

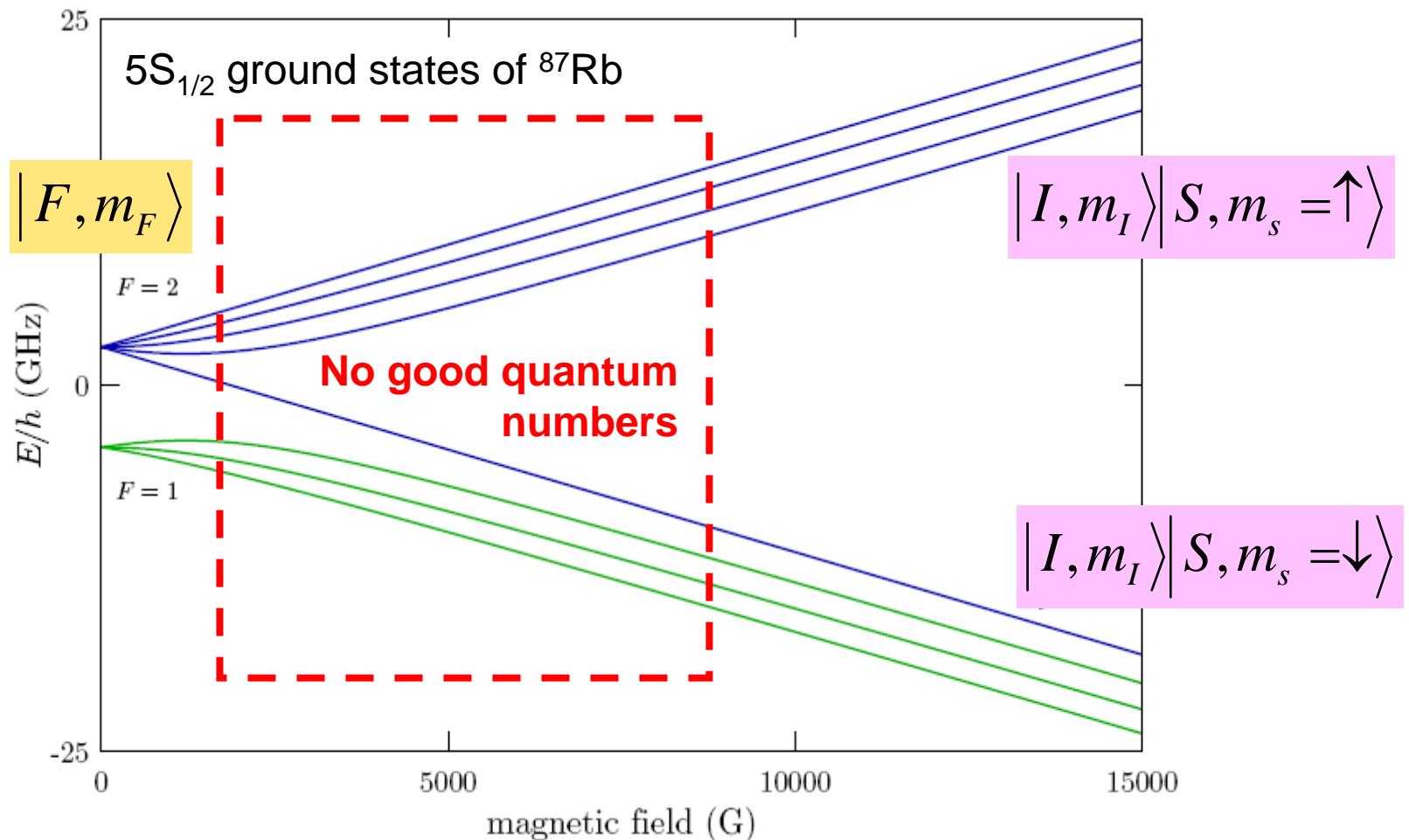
$$x \equiv \frac{(g_J - g_I) \mu_B B}{E_{hfs}}.$$

Zeeman Sub-Structure at High B-field



[Figure adapted from steck.us by Prof. Dan Steck, U. of Oregon (2010)]

Zeeman Sub-Structure at High B-field



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Clebsch-Gordan Coefficients: I=3/2, S=1/2

F=2 with I=3/2, S=1/2

$$|F = 2, m_F = +2\rangle = |I = 3/2, m_I = +3/2\rangle |\uparrow\rangle$$

$$|F = 2, m_F = +1\rangle = \frac{1}{2} |m_I = +3/2\rangle |\downarrow\rangle + \frac{\sqrt{3}}{2} |m_I = +1/2\rangle |\uparrow\rangle$$

$$|F = 2, m_F = 0\rangle = \frac{1}{\sqrt{2}} |m_I = +1/2\rangle |\downarrow\rangle + \frac{1}{\sqrt{2}} |m_I = -1/2\rangle |\uparrow\rangle$$

$$|F = 2, m_F = -1\rangle = \frac{1}{2} |m_I = -3/2\rangle |\uparrow\rangle + \frac{\sqrt{3}}{2} |m_I = -1/2\rangle |\downarrow\rangle$$

$$|F = 2, m_F = -2\rangle = |m_I = -3/2\rangle |\downarrow\rangle$$

Clebsch-Gordan Coefficients: I=3/2, S=1/2

F=1 with I=3/2, S=1/2

$$|F = 1, m_F = +1\rangle = \frac{\sqrt{3}}{2} |m_I = +3/2\rangle |\downarrow\rangle - \frac{1}{2} |m_I = +1/2\rangle |\uparrow\rangle$$

$$|F = 1, m_F = 0\rangle = \frac{1}{\sqrt{2}} |m_I = +1/2\rangle |\downarrow\rangle - \frac{1}{\sqrt{2}} |m_I = -1/2\rangle |\uparrow\rangle$$

$$|F = 2, m_F = -1\rangle = -\frac{\sqrt{3}}{2} |m_I = -3/2\rangle |\uparrow\rangle + \frac{1}{2} |m_I = -1/2\rangle |\downarrow\rangle$$

General Formula...

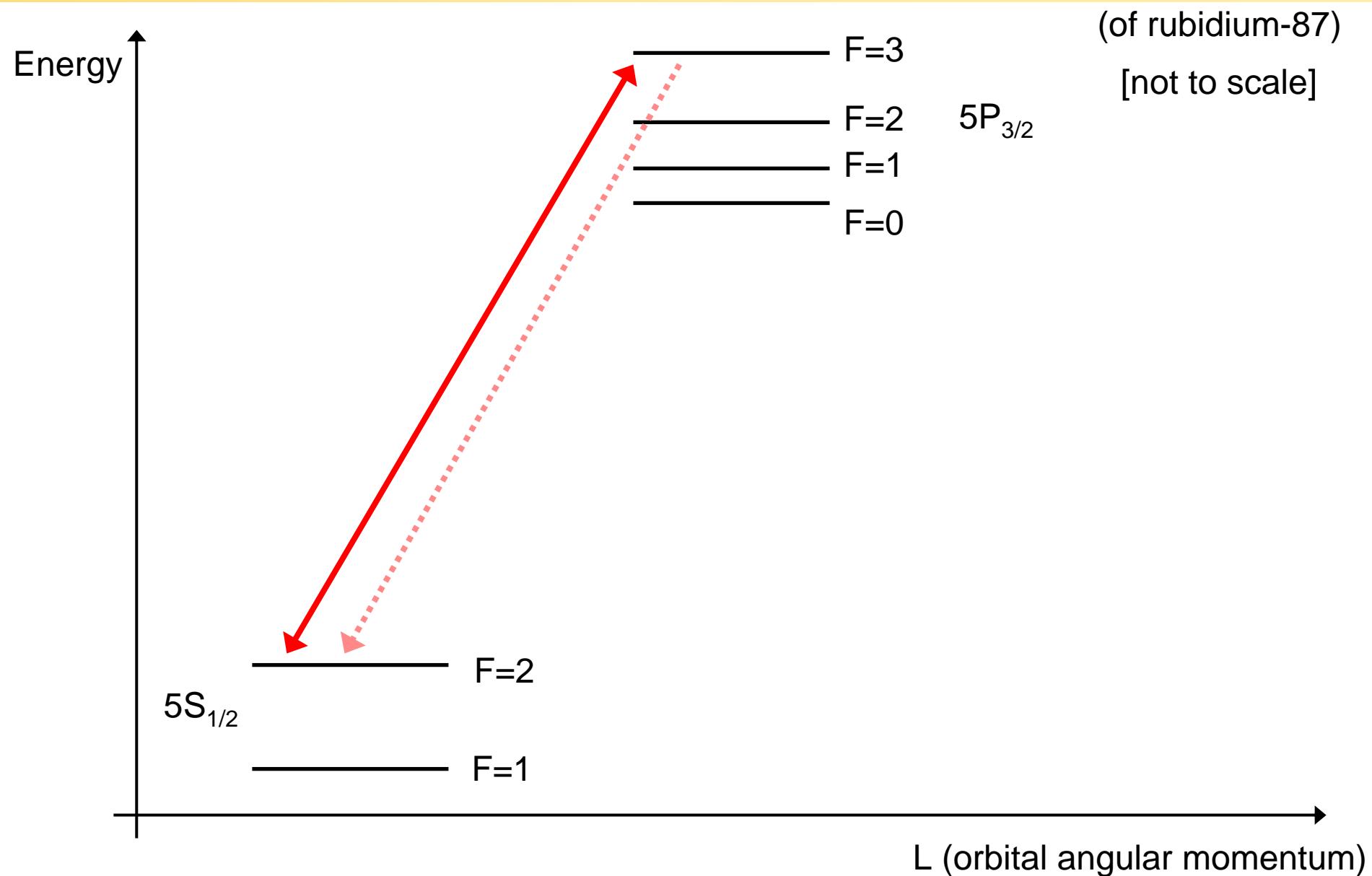
$$|F_+ = I + S, m_F \rangle =$$

$$\frac{\sqrt{F_+ + m_F}}{\sqrt{2I+1}} |m_I = m_F - 1/2 \rangle \uparrow \rangle + \frac{\sqrt{F_+ - m_F}}{\sqrt{2I+1}} |m_I = m_F + 1/2 \rangle \downarrow \rangle$$

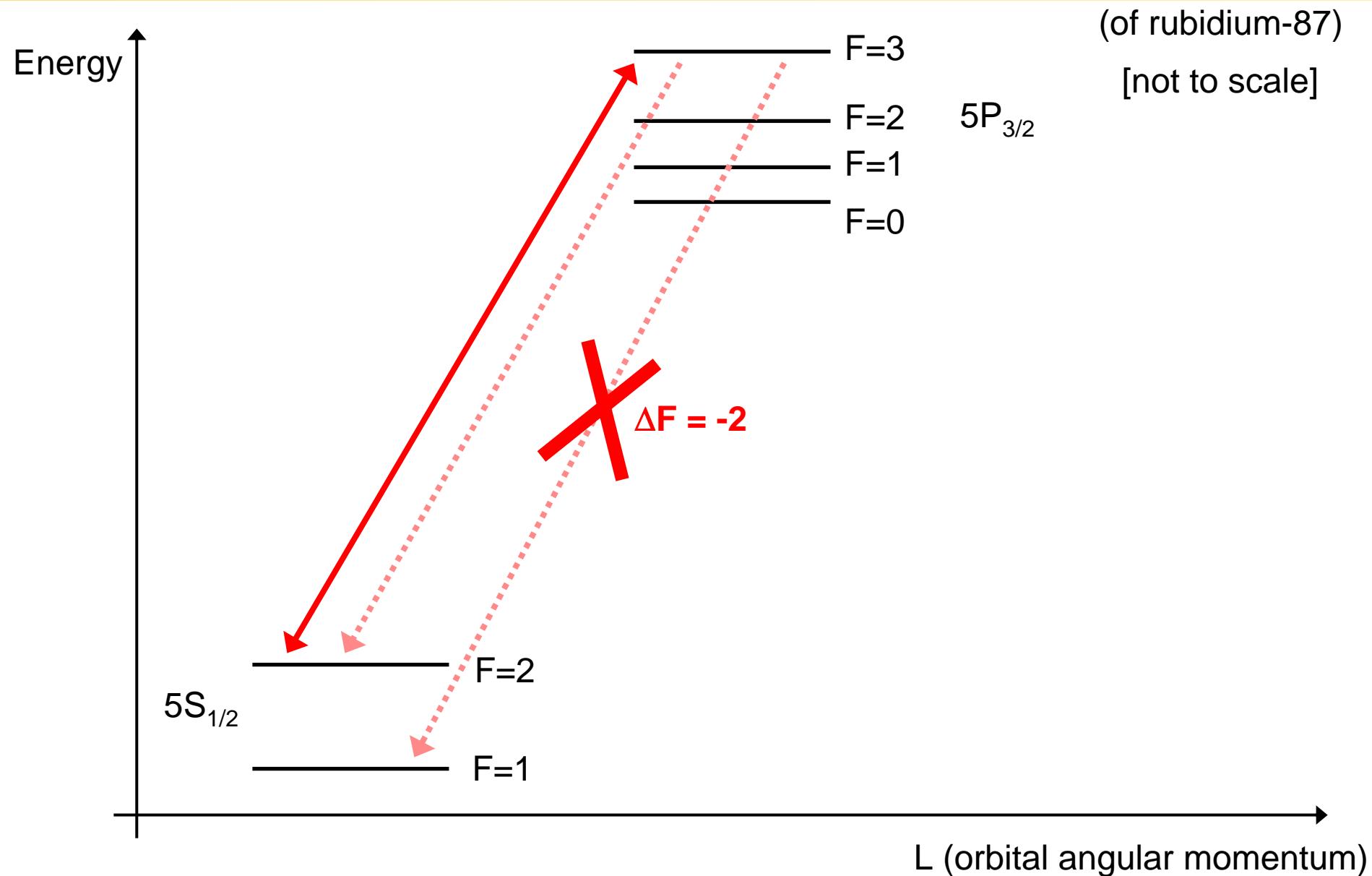
$$|F_- = I - S, m_F \rangle =$$

$$-\frac{\sqrt{F_+ - m_F}}{\sqrt{2I+1}} |m_I = m_F - 1/2 \rangle \uparrow \rangle + \frac{\sqrt{F_+ + m_F}}{\sqrt{2I+1}} |m_I = m_F + 1/2 \rangle \downarrow \rangle$$

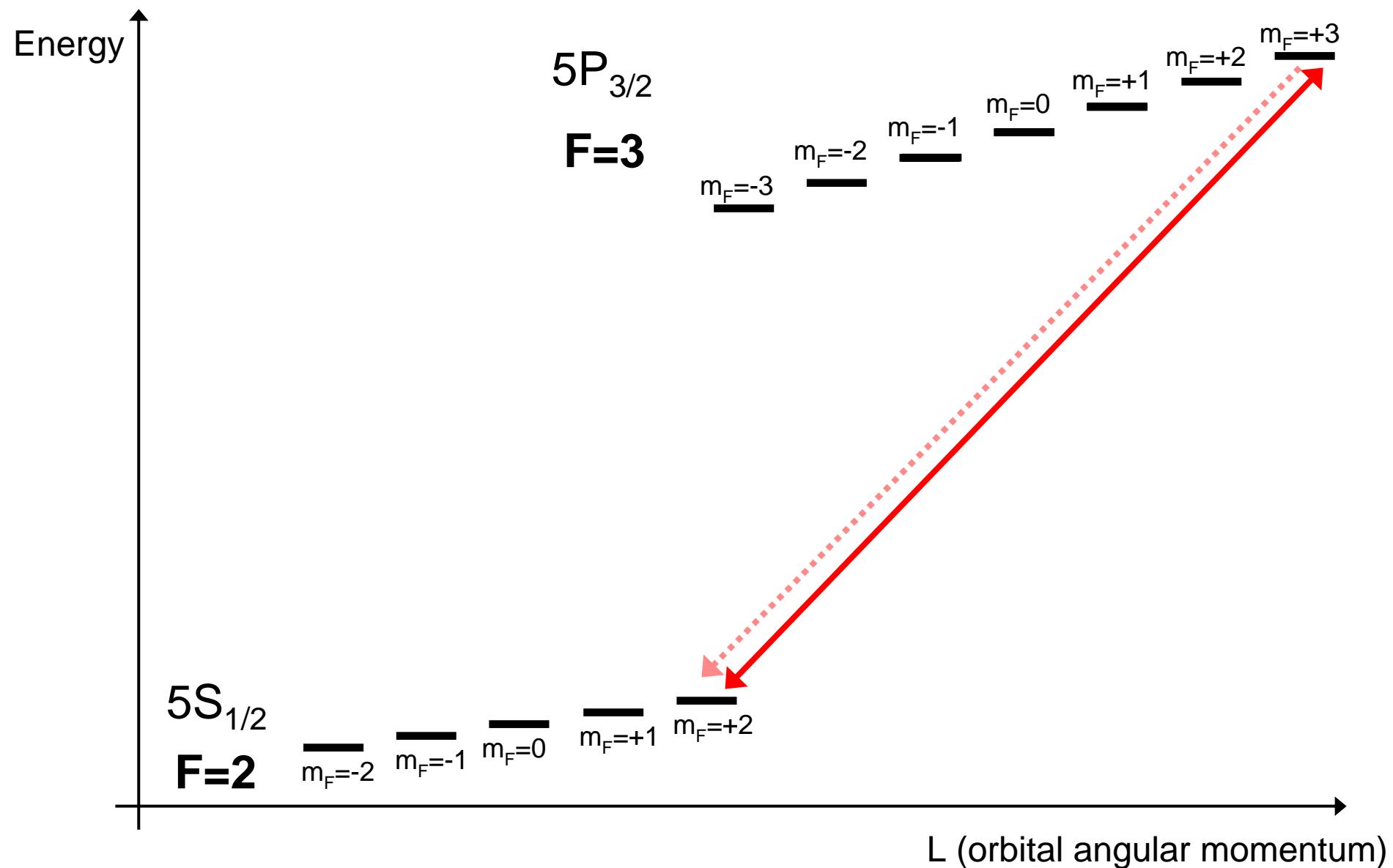
Why are Alkalies “2-level atoms” ?



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The D2 line Cycling Transition



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